Photo-rearrangement of o-Nitrobenzanilide

By B. C. GUNN and M. F. G. STEVENS*†

(Department of Pharmacy, Heriot-Watt University, Edinburgh EH1 2HJ)

Summary o-Nitrobenzanilide rearranges to 2-(2-hydroxyphenylazo)benzoic acid via an intermediate azoxybenzene derivative under the influence of light.

SUBSTITUENT interactions in *ortho*-substituted nitroarenes have been promoted thermally, photochemically, and by acids and bases;^{1,2} cyclised and uncyclised products are frequently encountered. The recently reported basecatalysed transformations of NN-disubstituted *o*-nitrobenzamides³ prompts us to report our results on the photolysis of *o*-nitrobenzanilide. Cream crystals of the anilide (1) turn maroon on exposure to ambient light, and an ethanol solution of the amide rapidly develops a blackcurrant colour when irradiated with an unfiltered 100w medium-pressure arc. The maroon photoproduct from the solution photolysis (15% yield after 12 h)[‡] was identical to that from the light-exposed crystals, and had a visible, i.r., n.m.r., and mass spectrum consistent with structure (3); in agreement, on catalytic hydrogenation it afforded only anthranilic acid and *o*-aminophenol. Although azo- and azoxy-compounds have been isolated from the photolysis of *ortho*-nitro-

† Present address: Department of Pharmacy, University of Aston in Birmingham, Gosta Green, Birmingham B4 7ET.
 ‡ Starting material (80%) was also recovered.

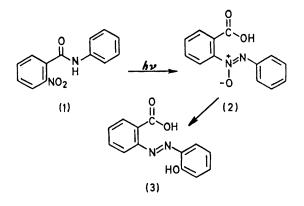
compounds, their formation in most cases can be attributed to bimolecular reactions involving reduction products.4 The formation of the azo-derivative (3), on the other hand, is unusual in that a unimolecular process seems to be involved.

We suggest that this isomerisation is initiated by abstraction of the amide N-H proton by the excited nitro-group,⁵ and proceeds to yield the azoxy-intermediate (2); further rearrangement of the azoxy-derivative to the final product (3) is in accord with the known photochemical azoxybenzene-hydroxyazobenzene rearrangement⁶ in which the oxygen atom always migrates to the ortho-position of the aryl group more distant from the N-O function.7 In support of this suggestion, the azoxybenzene (2) was detected (t.l.c.) in the photolysate, and a prepared sample⁸ was quantitatively converted into the hydroxyazobenzene isomer (3) in a separate photolysis.

This appears to be a general reaction of N-aryl-2-nitrobenzamides. For example, the 2-nitrobenzoyl derivative of α -naphthylamine rearranges to 2-(2-hydroxy-1-naphthylazo)benzoic acid in both crystal and solution phases. Of greater interest is the analogous rearrangement of 2-nitrobenz-o-toluidide, since this toluidide has been isolated as a metabolite of the currently-abused hypnotic, Metha-

- J. D. Loudon and G. Tennant, *Quart. Rev.*, 1964, 18, 398.
 P. N. Preston and G. Tennant, *Chem. Rev.*, in the press.
 T. W. M. Spence and G. Tennant, *J.C.S. Perkin I*, 1972, 835.

- ⁴ Y. Kitaura and T. Matsuura, *Tetrahedron*, 1971, 27, 1583.
 ⁵ H. A. Morrison, in 'The Chemistry of Nitro and Nitroso Groups,' Part 1, ed. H. Feuer, Interscience, New York, 1969, pp. 181–185.
- ⁶G. G. Spence, E. C. Taylor, and O. Buchardt, Chem. Rev., 1970, 70, 231.
- ⁷ G. M. Badger and R. G. Buttery, J. Chem. Soc., 1954, 2243.
 ⁸ A. Reissert and F. Lemmer, Ber., 1926, 59, 351.
- ⁹ T. Murata and I. Yamamoto, Chem. and Pharm. Bull. (Japan), 1970, 18, 133.



qualone.⁹ It is possible that the intense colour formed on photolysis of the toluidide could form the basis of a sensitive colour test to identify the Methaqualone metabolite in urine.

Financial support from S.R.C. (to B.C.G.) is gratefully acknowledged.

(Received, 18th May 1972; Com. 859.)